

## Acetyl Migrations in a Photoexcited $\beta\gamma$ -Unsaturated Methyl Ketone: 1,3-Shift in the Excited Singlet State and 1,2-Shift in the Triplet State<sup>1</sup>

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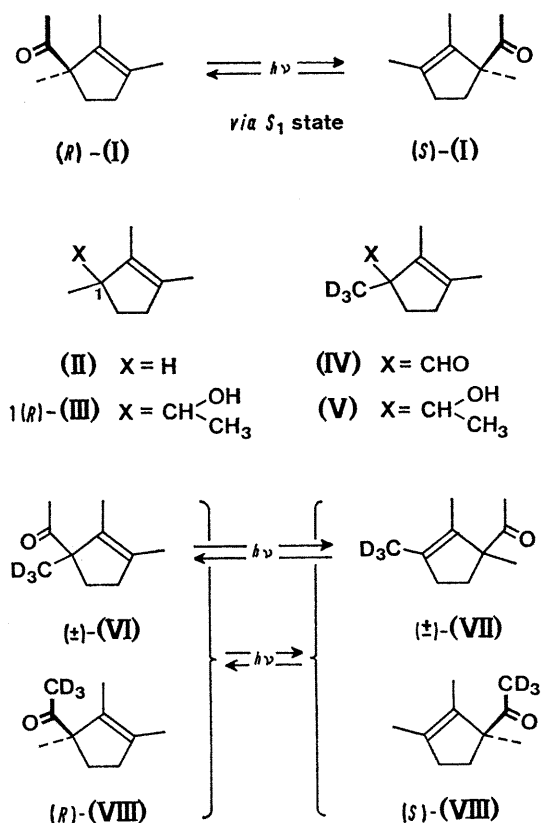
**Summary** In solution, 3-acetyl-1,2,3-trimethylcyclopent-1-ene (I) undergoes a sigmatropic 1,3-acetyl shift as the major unimolecular reaction in the excited singlet state, and a (thermatically reversible) 1,2-acetyl shift with cyclopropane ring-closure as the predominant triplet state reaction.

WE report on photochemical results obtained with (*R*)-3-acetyl-1,2,3-trimethylcyclopent-1-ene [(I);  $[\alpha]_D^{23} + 419^\circ$  (*c* 3.7,  $\text{CHCl}_3$ ); u.v. (iso-octane)  $\lambda_{\text{max}}$  300 nm. ( $\epsilon$  151); i.r. (film) 1705  $\text{cm}^{-1}$ ; n.m.r. ( $\text{CDCl}_3$ )  $\delta$  1.15 and 2.04 (two s, 3H each), 1.48 and 1.68 (two s with fine splitting, 3H each); 2,4-dinitrophenylhydrazone, m.p. 163°].<sup>2†</sup> Compound (*R*)-(I) was directly available from the reaction of (*R*)-lauroleonic acid<sup>3‡</sup> with 2 equiv. of methyl-lithium.<sup>4</sup>

U.v. irradiation of dilute solutions of the methyl ketone (*R*)-(I) in saturated hydrocarbons or benzene results in the formation of complex mixtures of low-yield products. Individual components have not yet been isolated except for one product whose composition and spectral data suggest the structure of a trimethylcyclopentenyl dimer. According to v.p.c. analysis, another of the components is 1,2,3-trimethylcyclopent-1-ene.<sup>7</sup> In the presence of tri-*n*-butyl stannane (1M in pentane) the proportion of this olefin was slightly increased in the photolysis mixture, and a new product, alcohol (III), was also formed in addition to the previous photoproducts. The rates of formation of these products, including (III), did not change significantly when 1M-naphthalene was added to a parallel run of (*R*)-(I) and stannane.

When the photolysis of a 0.13 M solution of methyl ketone (*R*)-(I) in pentane was interrupted after a consumption of 13% of ketone, the optical rotation of the recovered starting material had dropped from the initial value of  $[\alpha]_{436}^{23} + 1212^\circ$  (*c* 1.03) to  $+1071^\circ$  (*c* 1.45, both measurements in  $\text{CHCl}_3$ ). This photoracemisation accounts for over 32% of the detectable photochemical reaction of (*R*)-(I). That it is due entirely to a 1,3-acetyl migration was demonstrated by the following two photochemical experiments. The trideuteriomethyl compounds ( $\pm$ )-[<sup>2</sup>H<sub>3</sub>]-(*VI*), 100  $\pm$  1%, and (*R*)-[<sup>2</sup>H<sub>3</sub>]-(*VIII*), 93  $\pm$  1%, were prepared by methyl Grignard reaction of aldehyde (IV)<sup>5</sup> and oxidation<sup>8</sup> of the resulting two diastereoisomeric alcohols (V), and by hydrogen-deuterium exchange of (*R*)-(I) in tetrahydrofuran-D<sub>2</sub>O-NaOD, respectively. Irradiation of an equimolar mixture of ( $\pm$ )-(*VI*) and (*R*)-(*VIII*) (each 1.0 M in pentane) and mass spectral analysis of the recovered ketone mixture after interruption of the photolysis at 50% consumption of starting materials established the exclusive presence of [<sup>2</sup>H<sub>3</sub>]-ketones. Scrambling and formation of [<sup>2</sup>H<sub>6</sub>]-ketones

had not occurred. Furthermore, the rates of isomerisation of (*R*)-(I) and of (*VI*) were compared up to about 23% consumption of starting ketones. The conversion of (*VI*) into (*VII*) was measured by n.m.r. analysis of the appearance of the methyl signal at  $\delta$  1.15. Plots of percentage isomerisation against irradiation time for both runs gave straight lines with identical slopes in the range up to 13% ketone disappearance, and the rates were equal within  $\pm 1\%$  in the entire range.



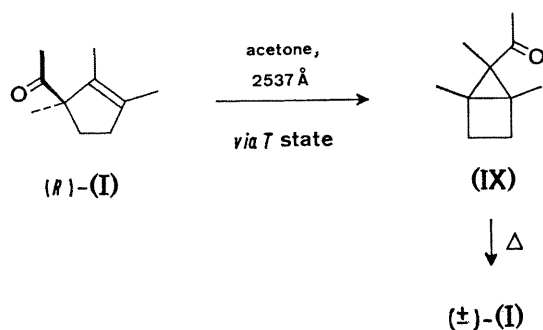
The rate of the photoracemisation of (*R*)-(I) was not affected upon the addition of triplet quenchers, 2M-penta-1,3-diene and 1 M-naphthalene. Alternatively, sensitisation of (*R*)-(I) in 0.05M acetone solution using 2537 Å radiation gave rise to a new photoisomer (IX) at the expense of both the photoracemisation and the formation of most of the product pattern observed on direct excitation of (*R*)-(I).

† Satisfactory elemental analyses were obtained for these two new compounds and for ketone (IX). The identification of the remaining new compounds, *i.e.* of alcohol (III) and the deuteriated substances (V), (VI), and (VIII), is based on spectral analysis (*i.e.*, n.m.r., and mass spectra) and on v.p.c. comparison of the deuteriated derivatives with the appropriate non-deuteriated analogues.

‡ The enantiomeric purity of the (*R*)-lauroleonic acid ( $[\alpha]_D^{23} + 185.4 \pm 1.1^\circ$ , liquid) which was used for the preparation of (*R*)-(I), is 96.3  $\pm$  1.4%.<sup>5</sup>

§ Sample solutions for direct irradiation of (I) were degassed by freeze-thaw cycling technique, sealed in Pyrex tubes and stirred magnetically at room temperature. A merry-go-round photoreactor<sup>6</sup> was used, equipped with a 125 W Hg medium pressure lamp which was surrounded by a filter absorbing all wavelengths below 327 nm.

The assignment of constitution (IXa), 5-acetyl-1,4,5-trimethylbicyclo[2,1,0]pentane, is consonant with the spectral data and the optical inactivity of the new product [u.v. (iso-octane)  $\lambda_{\max}$  300 nm. ( $\epsilon \sim 80$ ); i.r. (film) 1705  $\text{cm}^{-1}$ ; n.m.r. ( $\text{CDCl}_3$ )  $\delta$  1.15 (s with fine splitting, 3H), 1.17 (s, 6H), 1.5—2.0 (m, 4H), 2.21 (s, 3H)]. The mass spectrum of (IX) does not differ from that of (I) [prominent peaks at  $m/e$  152 ( $M^+$ , 1.6%), 109 ( $M^+ - \text{COCH}_3$ , 100%), 81 (17%), 67 (53%), 43 (43%)] except for some negligible differences in relative peak intensities and for the additional fragments  $M^+ - 15$  (3%),  $M^+ - 28$  (2.5%), and  $M^+ - 29$  (3%). The photolytic rearrangement (I)  $\rightarrow$  (IX) was reversed thermally at 200° in a v.p.c. SE-30 column. This thermal isomerisation of (IX) to (I) is novel for a bicyclo[2,1,0]pentanyl ketone, but has precedence in an analogous 1,2-ethoxy-carbonyl migration in structurally related ethyl (1,5-(di)-methylbicyclo[2,1,0]pentane-5-carboxylates.<sup>9\*\*</sup>



The results of the direct irradiation experiments with and without triplet quenchers, and of the sensitised photolysis in acetone are evidence that (i) singlet-triplet intersystem crossing in the  $\beta\gamma$ -unsaturated methylketone (I) is negligible, (ii) reaction modes of the excited singlet state are,

*inter alia*, 1,3-acetyl shift,  $\alpha$ -cleavage to acetyl and trimethylcyclopentenyl radicals, and reduction to carbinol (upon hydrogen abstraction from stannane), (iii) the predominant triplet reaction is a 1,2-acetyl shift and cyclopropane ring-closure. In several examples products of 1,2- and 1,3-acyl shifts in cyclic  $\beta\gamma$ -unsaturated ketones<sup>10</sup> have been either observed or have been invoked as the intermediates in apparent multi-step transformations. It has been generally assumed that these acyl migrations are preceded by  $\alpha$ -fission to allyl-acyl diradicals which then recombine to the appropriate  $\beta\gamma$ -unsaturated (starting or 1,3-rearranged compounds) or cyclopropyl ketones (1,2-rearrangement and ring closure). Compound (I) is the first  $\beta\gamma$ -unsaturated methyl ketone which has been found to photorearrange in a similar fashion. At the same time, however, the *aliphatic* nature of (I) sheds doubt on the occurrence of *free radical* intermediates in the rearrangement step. Furthermore, if radical primary products were produced in their electronic ground states, a specific choice between the two reaction paths by excited-state multiplicity of the reactant clearly would be unexpected. Our results suggest that both the 1,2- and the 1,3-acetyl migrations are concerted reaction steps, and that either one or both overall reactions are photochemically-allowed concerted [ $\sigma^2 + \pi^2$ ]-additions.<sup>11</sup> Subsequent to the recent preliminary presentation of our results,<sup>1</sup> Ipaktschi<sup>10b</sup> reported on similar specifically singlet- and triplet-initiated 1,3- and 1,2-acyl migrations in cyclic  $\beta\gamma$ -unsaturated ketones.

A further result of photochemical interest is, moreover, the reduction of the excited singlet state of ketone (I) to alcohol (III). A similar behaviour is observed with the related aldehyde, 3-formyl-1,2,3-trimethylcyclopent-1-ene, in the presence of tri-*n*-butylstannane.<sup>1,5</sup> In contrast to these findings, saturated as well as  $\alpha\beta$ -unsaturated and aryl ketones are known to reduce exclusively in their triplet ( $n, \pi^*$ ) state.<sup>12</sup>

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¶ Compound (I) does not exhibit phosphorescence emission at 78°K. in (ether:isopentane:alcohol) upon excitation in its  $n \rightarrow \pi^*$  absorption band.

\*\* The thermal isomerisation of 5-acetylbicyclo[2,1,0]pentanes analogous to (IX)  $\rightarrow$  (I) has been found independently (M. J. Jorgenson and A. F. Thacher, *Chem. Comm.*, 1969, 1030). We thank Dr. Jorgenson for this information, prior to publication.

<sup>1</sup> Presented in part at the meeting of the Swiss Chemical Society in Neuchâtel, February 15th, 1969; cf. E. Baggiolini, H. P. Hamlow, K. Schaffner, and O. Jeger, *Chimia*, 1969, **23**, 181. For previous paper in this series see R. Imhof, W. Graf, H. Wehrli, and K. Schaffner, *Chem. Comm.*, 1969, 852.

<sup>2</sup> C.d. data of methylketone (R)-(I) have been reported by G. Snatzke and K. Schaffner, *Helv. Chim. Acta*, 1968, **51**, 986.

<sup>3</sup> O. Aschan, *Ber.*, 1894, **27**, 2112, 3504; *Ann.*, 1896, **290**, 185.

<sup>4</sup> Method described by C. Tegner, *Acta Chim. Scand.*, 1952, **6**, 782, and references therein.

<sup>5</sup> E. Baggiolini, H. P. Hamlow, K. Schaffner, and O. Jeger, in preparation.

<sup>6</sup> G. S. Hammond, J. Saltiel, A. A. Lamola, N. J. Turro, J. S. Bradshaw, D. O. Cowan, R. C. Counsell, V. Vogt, and C. Dalton, *J. Amer. Chem. Soc.*, 1964, **86**, 3197.

<sup>7</sup> Preliminary results by A. Tuinman, Diploma Thesis ETH, Zurich 1966.

<sup>8</sup> Oxidation with sulphur trioxide-pyridine in dimethyl sulphoxide; cf. J. R. Parikh and W. v. E. Doering, *J. Amer. Chem. Soc.*, 1967, **89**, 5505.

<sup>9</sup> M. J. Jorgenson and T. J. Clark, *J. Amer. Chem. Soc.*, 1968, **90**, 2188.

<sup>10</sup> For the most recent reports and leading references see (a) R. K. Murray, jun. and H. Hart, *Tetrahedron Letters*, 1968, 4995; (b) J. Ipaktschi, *ibid.*, 1969, 2153; (c) K. Kojima, K. Sakai, and K. Tanabe, *ibid.*, p. 1925.

<sup>11</sup> R. Hoffmann and R. B. Woodward, *Accounts Chem. Res.*, 1968, **1**, 17.

<sup>12</sup> See, e.g., (a) P. J. Wagner, *J. Amer. Chem. Soc.*, 1967, **89**, 2503; P. Keller, G. Eggart, H. Wehrli, K. Schaffner, and O. Jeger *Helv. Chim. Acta*, 1967, **50**, 2259; R. Simonaitis, G. W. Cowell, and J. N. Pitts, jun., *Tetrahedron Letters*, 1967, 3751; (b) H. E. Zimmerman, R. G. Lewis, J. J. McCullough, A. Padwa, S. W. Staley, and M. Semmelhack, *J. Amer. Chem. Soc.*, 1966, **88**, 1965; (c) G. S. Hammond and P. A. Leermakers, *ibid.*, 1962, **84**, 207; J. A. Bell and H. Linschitz, *ibid.*, 1963, **85**, 528.